Dimensional Analysis Review Chem Worksheet 11-1

There are a variety of units that can be used when measuring. For example, the length of an object can be measured in millimeters, centimeters, meters, and even inches. A measurement made in inches can be converted to other units, such as centimeters using a conversion factor. A **conversion factor** is a ratio of two equivalent values expressed with different units. Name _____

Conversion Factors	
1 L = 1000 mL	1 km = 1000 m
1 m = 100 cm	$1 \text{ mm} = 1000 \ \mu \text{m}$
l gal = 3.785 L	1 in = 2.54 cm
1 km = 0.6214 mi	1 kg = 2.20 lb
1 yek = 18 mem	1 mem = 180 tezl

To solve conversion problems we use a strategy known as **dimensional analysis**. This technique focuses on canceling units by placing them on the top and bottom of the fractions used to set up a problem. For example, when converting 18 inches to centimeters we place the inches on the bottom of the conversion factor so that they cancel with the inches in the 'given'.

$\frac{18\,\text{m}}{1} \times \frac{2.54\,\text{cm}}{1\,\text{m}} = 45.72\,\text{cm}$



Convert the following measurements using dimensional analysis. Set up problem using fractions. Cross out the units that cancel. You must show work for credit.

- **1.** Convert 42.3 cm to m.
- 2. Convert the measurement 5.0 km to mi.
- **3**. Convert the measurement 150 lb to kg.
- 4. Convert 1.5 tezl to mem.
- **5.** Convert 2.00 liters to gal.

- 6. Convert 4.2 L to mL.
- 7. Convert the measurement 1.8 yek to mem.
- 8. Convert the measurement 325 mi to km.
- 9. Convert 180 cm to in.
- **10.** Convert 42 mem to yek.